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Low Pressure Chemical Vapor Deposition (LPCVD) Graphene Growth Study and Raman Characterization

by Barbara M. Nichols

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<p>Details of the Raman spectroscopy analysis of graphene films deposited for a growth study is discussed. Low pressure chemical vapor deposition was utilized to grow graphene layers onto copper foil substrates. For this study, the hydrogen-to-methane (H_2/CH_4) gas ratio used for the growths was varied from 20 to 200. Once transferred to patterned SiO_2/Si substrates, the graphene layers were characterized by Raman spectroscopy. Using the characteristic Raman peaks for graphene and silicon, maps of the integrated area ratio of the G and Si bands (A_G/A_{Si}), peak intensity ratio of the G' and G bands ($I_{G'}/I_G$), and the G' peak full-width-half-maximum (FWHM) were used to identify the number of layers and stacking orientation. Distinction between single-layer, Bernal-stacked bilayer, and turbostratic (or disoriented) bilayer graphene was made using this analysis. For the graphene deposited in this study, 100% monolayer graphene coverage was not achieved. All growths exhibited small areas of bilayer graphene. Although no correlation between the H_2/CH_4 gas ratio and the bilayer composition and/or stacking configuration was found, an increase in the size of the bilayer graphene was observed as the H_2/CH_4 gas ratio increased. This methodology is useful for relating graphene growth parameters to Raman properties.</p>					
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1. Introduction

Graphene is a monolayer material comprised of sp^2 bonded carbon atoms that form a honeycomb-like lattice structure. Because it is only one atom thick, it is the prototypical two-dimensional material and, as such, exhibits unique physical properties, including a high intrinsic mobility (200,000 cm^2/Vs) (1) and a constant optical absorption of 2.3% per layer over a wide spectral range (2). Graphene was first isolated physically by the mechanical exfoliation method (3); however, recent efforts have focused on graphene synthesis by conventional methods, such as chemical vapor deposition (CVD) and ultrahigh vacuum, high temperature annealing (i.e., epitaxial graphene from SiC) (4, 5). CVD, in particular, is a promising growth technique because of the ability to deposit large areas of graphene on inexpensive, transition metal materials (e.g., nickel and copper).

Graphene growth on nickel (Ni) and copper (Cu) metals exhibit two distinctly different mechanisms (6). At high temperatures (~ 1000 °C), carbon atoms dissolve into nickel up to ~ 1 atomic percent (at. %) (7). Then, as the temperature is lowered, the carbon atoms precipitate out of the nickel and form graphene on the Ni surface. Conversely, carbon only has a 0.001 at. % solubility in Cu and hardly any carbon atoms are dissolved in the metal (7). Graphene is formed by catalytic decomposition on the copper surface, and once the surface is covered, graphene formation is terminated. The different grown mechanisms generally explain why single-layer graphene (SLG) can be achieved on Cu; whereas, multilayer growth is typically found on Ni. Because of the desire to produce uniform single graphene layers, Cu has become the preferred substrate for CVD graphene growth.

Many parameters influence the growth of graphene on copper. These include pressure, gas flow rates, temperature, and the ratio between the carrier and carbon feedstock gases (i.e., hydrogen and methane) used for growth (8–10). In particular, hydrogen is known to play an important dual role in the growth of graphene by CVD methods. First, it acts as an activator for the surface bound carbon to form graphene; however, it can also serve as an etching agent, which helps control the size and shape of the graphene domains (8). CVD growth conditions have also been shown to determine the graphene grain size, which in turn have been correlated to electrical performance. Large grain graphene has produced higher mobility values (20,000 versus 1200 cm^2/Vs) and lower sheet resistances (20 versus 920 Ω/\square) than small-grained graphene on SiO_2 and hexagonal boron nitride (h-BN) (11). Larger grain graphene is typically grown at small hydrogen-to-methane (H_2/CH_4) ratio values (~ 1) and lower pressures (<500 mTorr) (11, 8); whereas, higher H_2/CH_4 ratio values (e.g., 40) tend to promote the formation of bilayer graphene (9).

In this report, we discuss the Raman spectroscopy results of a study varying the H_2/CH_4 gas flow ratio for graphene grown by low pressure chemical vapor deposition (LPCVD). First, the

graphene was grown on copper foils and transferred onto grid-patterned SiO₂/Si substrates. After transfer, the samples were characterized by Raman spectroscopy. Analysis of the Raman data showed the coexistence of both single-layer and double-layer (or bilayer) graphene over areas as large as 80 × 80 μm. Details of the analysis will be given as well as a discussion of the Raman characteristics that distinguish ordered (AB or Bernal-stacked) and disoriented (or turbostratic) bilayer graphene (BLG).

2. Experimental Procedure

Graphene was synthesized by LPCVD using hydrogen and methane as the carrier and carbon feedstock gases. Copper foils (99.8% pure), approximately 25 and 125 μm thick, were cleaned with acetic acid, acetone, and isopropanol, and loaded into the furnace. The furnace was pumped down to a base pressure less than 0.5 Torr and then heated to 1000 °C in 100 sccm of H₂. Once the furnace reached temperature, the copper foils were annealed for 30 min under flowing H₂. Next, CH₄ was metered into the furnace with the throttle valve open 100% to the mechanical vacuum pump. Growth pressure ranged between 0.5 and 0.6 Torr. After 20 min, the methane flow was stopped, and the furnace was cooled to room temperature under 100 sccm of H₂. For these experiments, the H₂/CH₄ gas flow ratio was varied by keeping the hydrogen flow fixed at 100 sccm and changing the methane flow rate. The methane flow rate was varied from 0.5 to 5 sccm. A summary of the growth conditions can be seen in table 1. After growth, the samples were examined with a scanning electron microscope (SEM).

Table 1. Summary of the LPCVD parameters for the H₂/CH₄ graphene growth study.

Sample	H ₂ Flow (sccm)	CH ₄ Flow (sccm)	H ₂ /CH ₄ Ratio	Time (min)	Pressure (Torr)	Temperature (°C)
LPC065	100	5	20	20	0.56	1000
LPC066	100	2	50	20	0.55	1000
LPC067	100	3	33	20	0.59	1000
LPC068	100	1	100	20	0.55	1000
LPC069	100	0.5	200	20	0.55	1000
LPC070	100	4	25	20	0.56	1000

After growth, the graphene was transferred to patterned SiO₂/Si substrates via a wet chemical etching process of the copper foil. First, a protective layer of poly(methyl methacrylate) (PMMA) was spun onto the graphene using either 495 PMMA A4 (4% PMMA in anisole), 495 PMMA A8 (8% PMMA in anisole), or a combination of the two and baked at 150 °C for 90 s. Next, the unprotected side of the copper foil was exposed to an oxygen plasma (power = 50 W, O₂ flow = 200 sccm, pressure = 0.5 Torr) to remove the backside graphene layer. The copper foil was then chemically etched away using a commercial ferric chloride based etchant (Transene CE-100), followed by a 10% HCl in H₂O etch to remove any particulate residue from the copper

etchant. After each etch step, the PMMA/graphene was floated on water baths to immediately stop any chemical reactions. The floating PMMA/graphene was then placed onto the substrate and baked on a hot plate between 50 and 75 °C for 7–10 min to drive away excess water between the substrate and the PMMA/graphene layer. The patterned substrates were designed and fabricated at the U.S. Army Research Laboratory (ARL) as a marker grid to locate areas on the substrate. The grid structure was patterned using standard photolithography techniques and buffer oxide etched into the silicon dioxide approximately 100–150 nm deep (see figure 1). The grid was visible underneath the graphene. Removal of the PMMA layer was performed by soaking the sample in chloroform for 90 min, followed by an isopropanol/acetone/isopropanol rinsing. Prior to the chloroform soaking, the sample was baked on a hot plate (~200 °C) for 5 min to dewrinkle/smooth the PMMA/graphene on the substrate. The samples were later thermal annealed at 350 °C for 2 h in a mixed hydrogen/argon atmosphere to further remove any PMMA residue found on the surface.

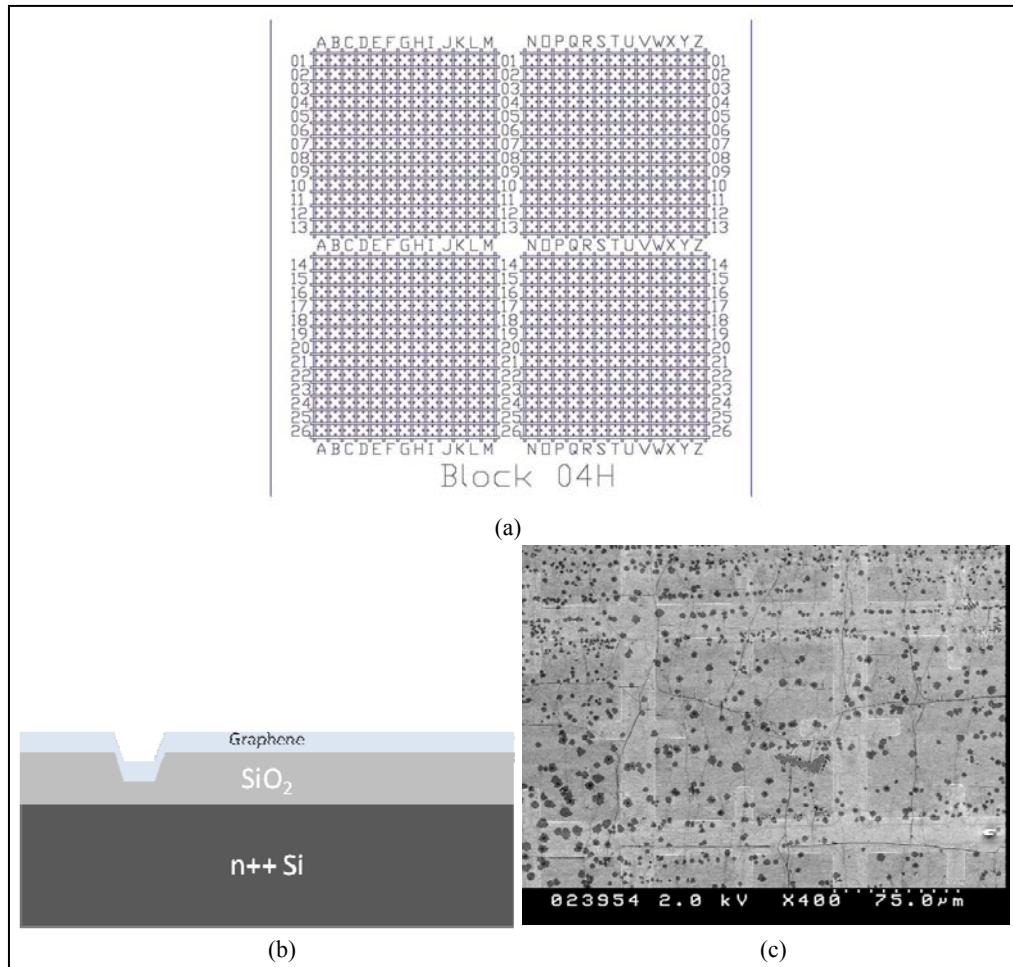


Figure 1. (a) Drawing of the grid pattern etched into SiO₂/Si substrates for identification purposes, (b) Cross-sectional drawing of the transferred graphene samples, and (c) a SEM image of graphene transferred onto a patterned substrate (LPC068B1TCA – H₂ = 100 sccm, CH₄ = 1 sccm).

Raman spectroscopy measurements were performed to characterize the graphene after transfer, after PMMA removal via chloroform, and after thermal annealing. A WITec alpha300RA confocal Raman microscope (CRM) allowed for the acquisition of individual Raman spectrum over large areas (up to $80 \times 80 \mu\text{m}$) with submicron spacing. The CRM was operated with a laser wavelength of 532 nm, ~1.5 mW power at the sample, and a 600 grooves/mm grating. This capability produced detailed views of the transferred films that revealed information about the number of graphene layers and other features, such as wrinkles and folds. Because of the patterned silicon dioxide substrates, the exact graphene location was scanned after the three steps mentioned above to specifically detail if/how the PMMA removal steps affected the graphene. The Raman spectroscopy data was analyzed using the Lorentz peak fitting approximation for the

characteristic graphene peaks – G' peak at \sim 2670 cm $^{-1}$, G peak at \sim 1580 cm $^{-1}$ and D peak at \sim 1350 cm $^{-1}$, as well as the Si peak at \sim 520 cm $^{-1}$. Once fitted, the individual peak features, such as full-width-half-maximum (FWHM), peak height, and peak position, could be plotted as maps, revealing structural information about the graphene, including the number of layers and structural stacking configurations.

3. Results and Discussion

3.1 Raman Characterization

The results of a Raman mapping scan of one transferred graphene growth can be seen in figure 2. Figure 2a shows a $100 \times 100 \mu\text{m}$ optical image (taken under a green filter) of a transferred graphene film (LPC069B1TC). The optical image shows a large monochromatic field with multiple darker patches and lines. The four lighter rectangles in the figure are the etched SiO₂ trenches, which are covered by the graphene film. Using the maps developed from the Raman spectra (1 individual spectra was collected per pixel), one can correlate features in the optical image with those observed in the Raman maps. The location of bright areas seen in the $A_{G'}/A_{Si}$ (peak area ratio between the G and Si peaks) map in figure 2b correspond to the dark patches in the optical image, revealing that the bright locations are composed of two or more graphene layers. Aside from the etched trenches, dark areas indicate the presence of no graphene, likely due to a slight rip or tear in the otherwise continuous graphene film. A map of the G' peak FWHM can be found in figure 2c. It has been shown that the graphene G' peak is extremely sensitive to both the number and stacking of bilayer and few layer graphene (12). Wider G' peaks have been shown to belong to energetically favored stacking in multiple graphene layered materials, such as AB (Bernal-stacked) bilayer or ABC/ABA trilayer; whereas, turbostratically oriented layers (layers randomly rotated with respect to one another) typically have narrower widths than single-layer graphene (9). In figure 2c, the areas composed of two or more layers have FWHM values that are either smaller ($28\text{--}32 \text{ cm}^{-1}$) or larger ($>45 \text{ cm}^{-1}$) than the single-layer median value of 36 cm^{-1} . The third important feature for graphene is the peak height or intensity ratio between the G' and G peaks, denoted as $I_{G'}/I_G$ (see figure 2d). Early on, the peak intensity ratio $I_{G'}/I_G$ was used as an indicator of graphene layer number and with recent understanding, it can also be used to infer stacking order (9). The map has a median intensity ratio of 1.75, which is near but less than the $I_{G'}/I_G$ value of 2, which has been used as an indicator of single-layer graphene. However, the map also reveals areas with significantly higher (~ 2.6) and lower (~ 0.8) values than the median. Using these three variable maps, $A_{G'}/A_{Si}$, G' FWHM, and $I_{G'}/I_G$, one can differentiate areas of the graphene by both layer count and stacking configuration, which is particularly important because the electrical properties of graphene strongly depend upon these variables.

Figure 3 shows a map identifying the graphene areas observed in figure 2. An approach similar to reference 13 was taken to develop the map. The majority of the scanned area is composed of single-layer graphene; however, both Bernal-stacked and turbostratic bilayer areas have been identified. The graphene areas over the etched trenches were excluded from the compositional identification because the low Raman spectra intensity at these areas made it difficult to perform peak fitting (and thus are colored gray in figure 3a). The black areas in the image are locations where the composition could not be determined because they did not fit the general guidelines discussed above. These locations are primarily composed of the wrinkles/folds observed in the image. Because the folds/wrinkles are formed as a result of thermal expansion differences between the graphene and copper during the cool down phase of the growth or from the transfer process, it is expected that they could exhibit properties different from AB-stacked or turbostratic BLG (4). There are also bilayer areas where the stacking configuration could not be determined. These areas most likely have G'FWHM or I_G/I_G values that typically mark transition areas between different stacking configurations or layers. While every pixel of the map was not able to be definitively identified, a large majority were identified and the results can be used to give a composition summary of the scanned area. Based on the analysis, the scan area found in figures 2 and 3 is composed of 88% SLG, 3% turbostratic BLG (T-BLG) and 1.9% Bernal-stacked BLG (AB-BLG). The analysis also indicates that 0.5% of the scanned area is composed on trilayer graphene. Wrinkles and other unidentifiable areas make up 6.5%.

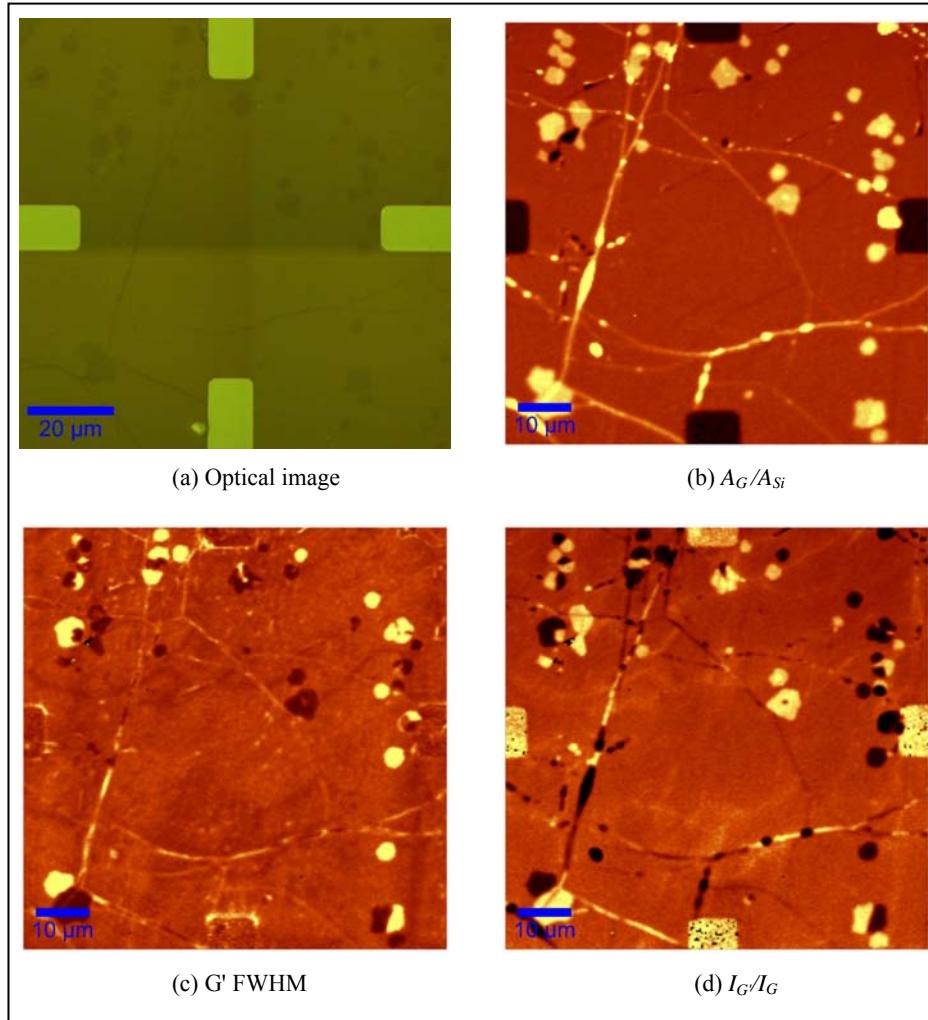


Figure 2. Optical and Raman images of a graphene film transferred onto patterned oxide substrate (LPC069B1TC – $H_2 = 100$ sccm, $CH_4 = 0.5$ sccm). (a) Optical image of the scanned area, Raman maps of the (b) peak area ratio A_G/A_{Si} , (c) G' FWHM, and (d) the intensity peak ratio $I_{G'}/I_G$ of the same area shown in the optical image.

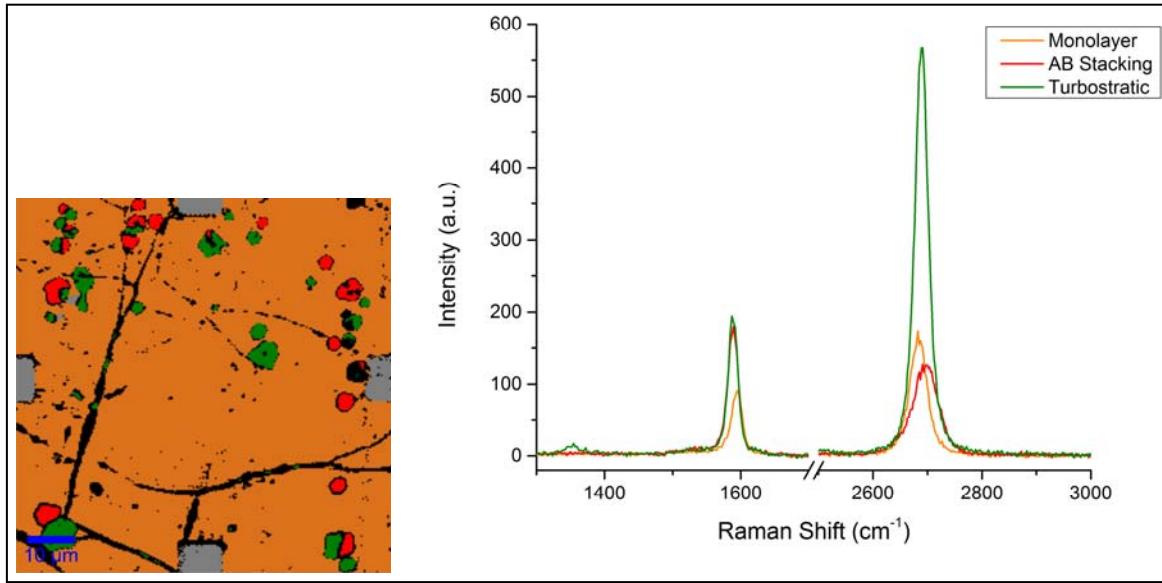


Figure 3. (a) Composite map of LPC069B1TC showing the composition of the Raman maps shown in figure 2 and (b) the corresponding Raman spectra for the three different graphene configurations—orange (SLG), red (AB-stacked bilayer), and green (turbostratic bilayer).

Representative Raman spectra for the three major compositions (SLG, AB-BLG, and T-BLG) are shown in figure 3b. The key features of the Raman spectra discussed earlier (e.g., the increased FWHM for Bernal-stacked material and differences in the G and G' peak heights) yield very distinctive spectra for each composition. Peak position shifting can also be seen for the three different configurations. A summary of the median peak position and FWHM values can be found in table 2. As expected, the G' peak position shows the most variability. The bilayer compositions show significant blue shifts from the median SLG position—7 cm⁻¹ and 10.5 cm⁻¹ for turbostratic and bernal stacking, respectively. The bilayer G peak position shows a smaller shift from SLG in the opposite direction (3.6–4.6 cm⁻¹ red shift). Similar shifts in the position of G and G' peak have been observed in exfoliated turbostratic and AB-BLG (12, 14).

Table 2. Summary of the peak fitting median values for single-layer, Bernal-stacked bilayer, and turbostratic bilayer graphene for LPC069B1TC (as seen in figure 3).

	G' Position	G' FWHM	G Position	G FWHM	I _{G'} /I _G	A _G /A _{Si}
Whole Image	2683.7	35.7	1592.4	17.0	1.8	0.14
SLG	2682.5	35.7	1592.6	17.0	1.8	0.14
AB-BLG	2693.0	56.3	1588.0	17.8	0.8	0.28
T-BLG	2689.5	29.0	1589.0	16.0	2.8	0.25

3.2 H₂/CH₄ Growth Study Results

For the films grown in this study, the $I_{G'}/I_G$ intensity map for each growth can be seen in figure 4. Two main features are observable in every Raman image. The first is the presence of folds in the graphene. It is thought that the folds more likely originate during the cool down phase of graphene growth than during the transfer process. As discussed earlier, they tend to have Raman peak characteristics that veer slightly from typical SLG and BLG values. Secondly, all images show the growths are mostly monolayer graphene with areas of bilayer graphene present. The composition of the scanned areas for each growth can be found in table 3. The values range from 83% to 93% SLG. The amounts of turbostratic and Bernal BLG varied, as well with no direct correlation to the methane flow rate. However, the size of the bilayer regions did appear to correlate with the H₂/CH₄ ratio. The average bilayer region ranged from 1 to 4 μm as the methane flow decreased (or as H₂/CH₄ increased). These results are very similar to those performed previously, where the CH₄ flow was held at 5 sccm and the H₂ flow rate was varied from 500 to 0 sccm (15). This suggests that the hydrogen-to-methane flow ratio plays an important role in the formation of bilayer graphene and understanding this relationship provides a pathway for producing large coverage bilayer graphene. Also noteworthy in figure 4 is the stacking of individual bilayer areas. It seems that one bilayer area or “grain” can consist of two or more different stacking configurations. For example, in the A_G/A_{Si} map of figure 2, one larger bilayer grain can be seen in the lower right hand corner and appears as one continuous area. However, both the G' FWHM and $I_{G'}/I_G$ maps show this area to have two different stacking configurations. In exfoliated graphene, two or more stacking configurations within a trilayer graphene flake has been routinely observed (16). However, this is not the case for exfoliated bilayer graphene. Having multiple stacking configurations in BLG is unique to CVD graphene. At this point, it is unclear as to what causes this phenomena.

Table 3. Composition of the graphene films for the H₂/CH₄ growth study based on the analysis of the Raman scanned areas.

Sample	T-BLG	AB-BLG	SLG	H ₂	CH ₄	H ₂ /CH ₄
069B1TC	3.0%	1.9%	88.0%	100	0.5	200
068B1TC	3.0%	4.9%	82.5%	100	1	100
066B1TC	2.6%	3.7%	86.3%	100	2	50
067B1TC	1.9%	3.6%	85.2%	100	3	33
070B1TC	2.9%	3.7%	86.4%	100	4	25
065B1TC	1.6%	1.0%	93.0%	100	5	20

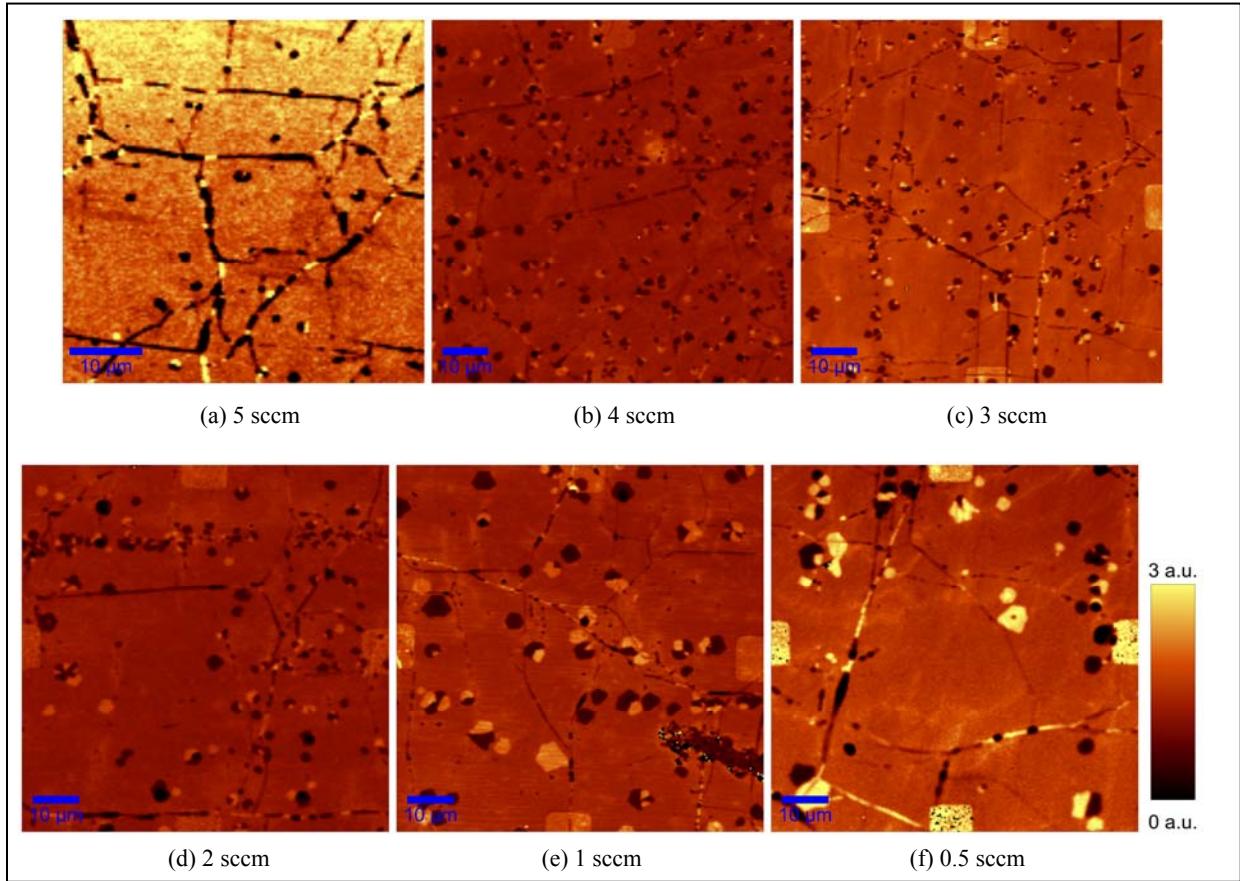


Figure 4. I_G/I_G peak intensity maps for the six graphene films grown with different CH₄ flow rates. The color scale shown in the figure was used for all six maps.

The quality of the graphene is typically examined by looking at the characteristic D peak. This peak arises due to symmetry breaking defects present in the graphene and is thought of as a measure for disorder with the sp² carbon matrix. A comparison of the D peak intensities for the growth series can be seen in figure 5. Although the intensities are quite low when compared to the G peak intensity ($I_D/I_G < 0.1$) and therefore considered good quality, correlations between areas of high D intensities and the I_G/I_G maps (see figure 4) can be observed. The major folds and wrinkles found in the graphene typically exhibit higher intensities of the D peak. However, most strikingly, areas that exhibit high D peak intensities are the areas that were previously identified as turbostratic or disoriented bilayer graphene. Because turbostratic BLG does not exhibit the low-energy stacking configuration, it does seem that two layers randomly oriented or rotated by one another would exhibit the symmetry breaking properties that typically give rise to the characteristic D peak. Based on this observation, it seems that the quality of the graphene should not be based solely on the presence of the D peak in the Raman spectra.

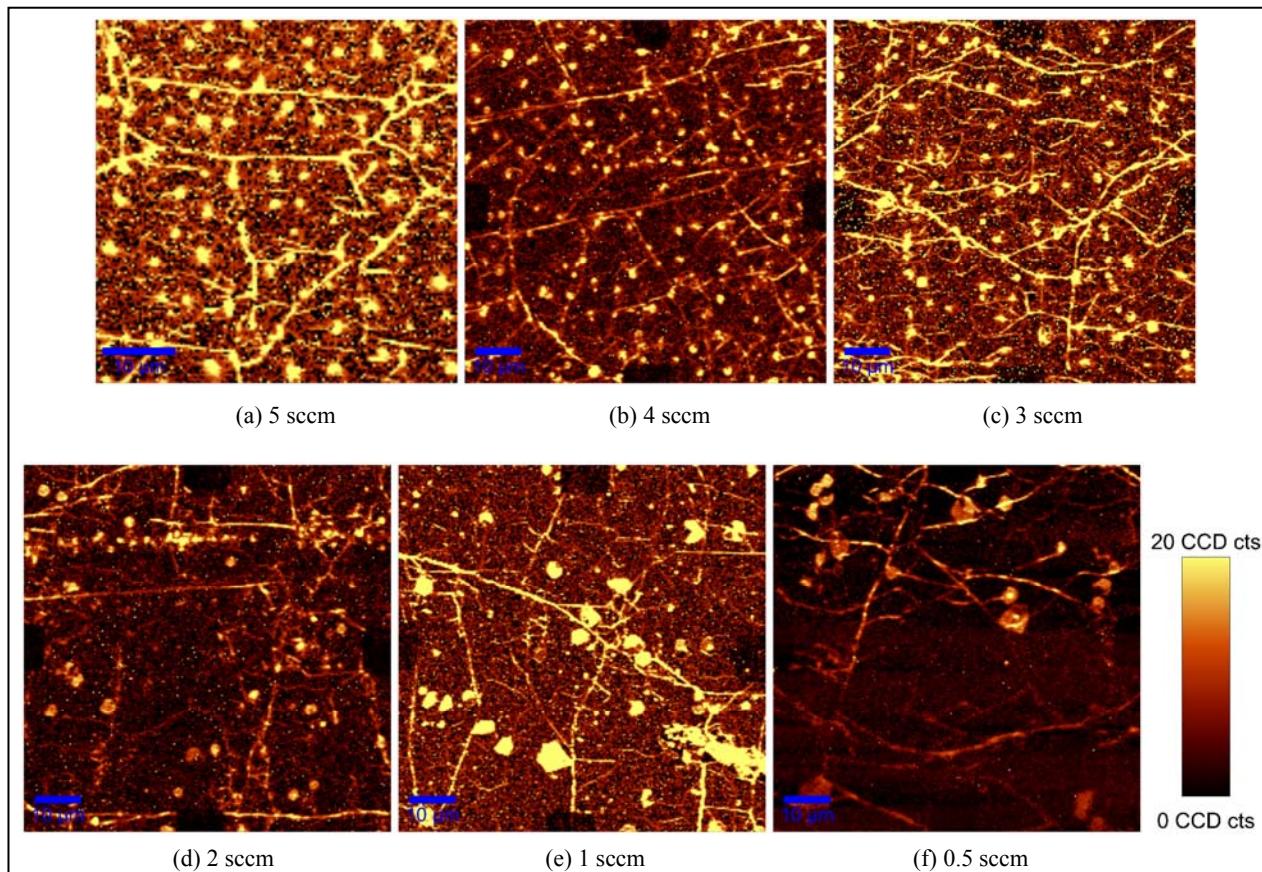


Figure 5. D peak intensity maps for the six graphene films grown under different CH₄ flow rates. The color scale shown in the figure was used for all six maps.

4. Conclusions

Raman characterization of CVD graphene grown under varying H₂/CH₄ flow ratio values was explored in this report. The growth study reveals that although the H₂/CH₄ ratio varied from 20 to 200, 100% single-layer graphene was not achieved. All growths were composed primarily of SLG with noticeable amounts of bilayer graphene present. The percentage of BLG in the graphene did not correlate with the flow ratio. However, the size of the BLG areas in each scan increased as the H₂/CH₄ increased. It should also be noted the characteristic D peak, a measure of the symmetry breaking in the sp² lattice, correlated to the presence of turbostratic or disoriented BLG graphene and not Bernal BLG. This analysis has proven useful to correlate growth parameters to the resulting graphene structural properties.

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List of Symbols, Abbreviations, and Acronyms

AB-BLG	Bernal-stacked BLG
ARL	U.S. Army Research Laboratory
BLG	bilayer graphene
CRM	confocal Raman microscope
Cu	copper
CVD	chemical vapor deposition
FWHM	full-width–half-maximum
h-BN	hexagonal boron nitride
LPCVD	Low-Pressure Chemical Vapor Deposition
Ni	nickel
PMMA	poly(methyl methacrylate)
SEM	scanning electron microscope
SLG	single-layer graphene
T-BLG	turbostratic BLG

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